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Determine the pH of a solution made by dissolving 4.50 g NaOH in a 0.400 L aqueous solution. NaOH is a strong base. Solution ANALYZE What is given in the problem? the mass of NaOH, and the solution volume What are you asked to find? pH PLAN What steps are needed to calculate the pH? First determine the concentration of the solution. Then find the concentration

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write an equation linking the term pH with the term [H3O+] Indicators rely on color changes certain dyes undergo as pH changes and can be used to roughly measure pH within a range. pH meters measure the voltage between two electrodes and convert it to a precise pH reading

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